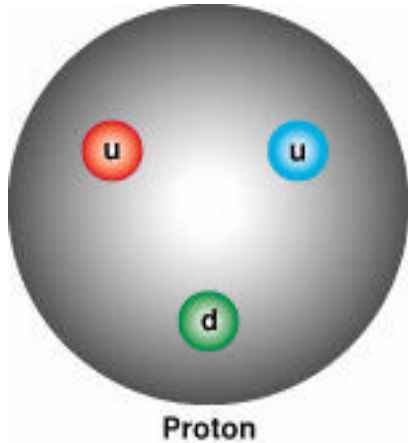


The nucleus



- ✓ Notation
- ✓ Forces within the nucleus
- ✓ Binding energy of nuclei
- ✓ The quantum mechanical system
- ✓ Angular momentum
- ✓ Parity
- ✓ Multipoles
- ✓ Radioactivity
- ✓ alpha, beta decays
- ✓ isomeric transitions (gamma decays)

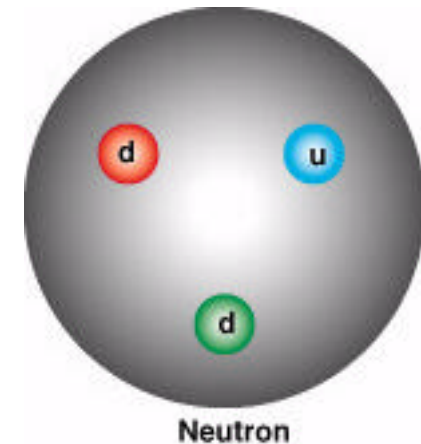
Notation and terminology



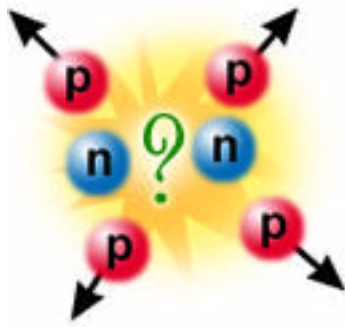
- ✓ nucleus = Z protons + N neutrons = A nucleons
- ✓ nuclide = nuclear species = same Z and N
- ✓ isotopes = nuclides with same Z , different N

✓ isotope notation: ${}^A_Z X_N$ or ${}^A X$

- ✓ isobars = nuclides with the same A
- ✓ isotones = nuclides with the same N



Forces in the nucleus



- ✓ strong interaction: attractive nuclear force between nucleons
- ✓ similarity between p and n
- ✓ size of nuclei ~ 1 fm
- ✓ electromagnetic interaction: repulsion between protons
- ✓ no analytical expression for nuclear forces: models

Binding energy

$B = (\text{mass of nucleons}) - (\text{mass of nucleus})$

$$\text{e.g. } m_n + m_p - m_D = B > 0$$

the potential energy of the nucleus is negative

$$m_n + m_p + (-|E_{int}|) = m_D \quad \text{so} \quad B = |E_{int}|$$

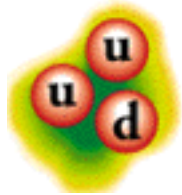
the binding energy per nucleon B/A varies with A
and is in the interval 5 MeV - 10 MeV (Krane)

- ✓ $A > 62$: B/A decreases with A : fission
- ✓ $A < 62$: B/A increases with A : fusion

The nucleus as a quantum mechanical system

- nuclear states: energy, L, S, P eigenstates
- nuclear energies: constant , quantized
- nuclei have a ground state (possibly excited states)
- transitions between states:
 - alpha particles (alpha decay)
 - electrons or positrons (beta decay)
 - photons (gamma- or X-rays)

Angular momentum and parity (J^+ or J^-)



- ✓ protons and neutrons have spin $s=1/2$
- ✓ protons and neutrons move inside the nucleus and have orbital angular momentum l

How do we add the l 's and s 's of the nucleons?

$$J = \sum j_{\text{nucleon}} \text{ 'angular momentum' or 'spin'}$$

$$\text{parity of the nucleus: } \lambda = (-1)^l$$

where l is the total orbital angular momentum of the nucleus

Nuclear e/m moments (Krane §3.5)

Electric multipole moments:

- 0th or monopole moment: the electric field varies as r^{-2} (L=0)
- 1st or dipole moment: the electric field varies as r^{-3} (L=1)
- 2nd or quadrupole moment: the electric field varies as r^{-4} (L=2), etc.

where L is the order of the moment

We have the same for magnetic multipole moments except that there is no monopole moment

$$\begin{aligned}\text{Parity of electric moments} &= (-1)^L \\ \text{Parity of magnetic moments} &= (-1)^{L+1}\end{aligned}$$

Radioactivity

‘spontaneous transition from one nuclear state to another’

During the transition: quantum numbers may change

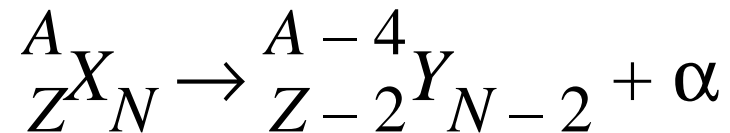
In alpha and beta decays: both Z and N change

- ✓ exponential decay law: $N = N_0 e^{-\lambda t}$
- ✓ activity of the sample: number of decays per unit time
- ✓ unit for activity: 1 becquerel, Bq = 1 decay per second
- ✓ decay constant, λ : probability for decay per time unit
- ✓ half-time: $t_{1/2} = \ln 2 / \lambda$
- ✓ lifetime: $\tau = 1 / \lambda$

Conservation rules in nuclear decays

- ✓ Baryon and lepton number always conserved
- ✓ Parity of the nucleus may be violated in a weak decay
- ✓ Total energy, total angular momentum (orbital+spin), total linear momentum are always conserved

Alpha decay



Q value of the decay:

$$Q = m({}^A X) - m({}^{A-4} Y) - m({}^4 He) \quad (\text{atomic masses})$$

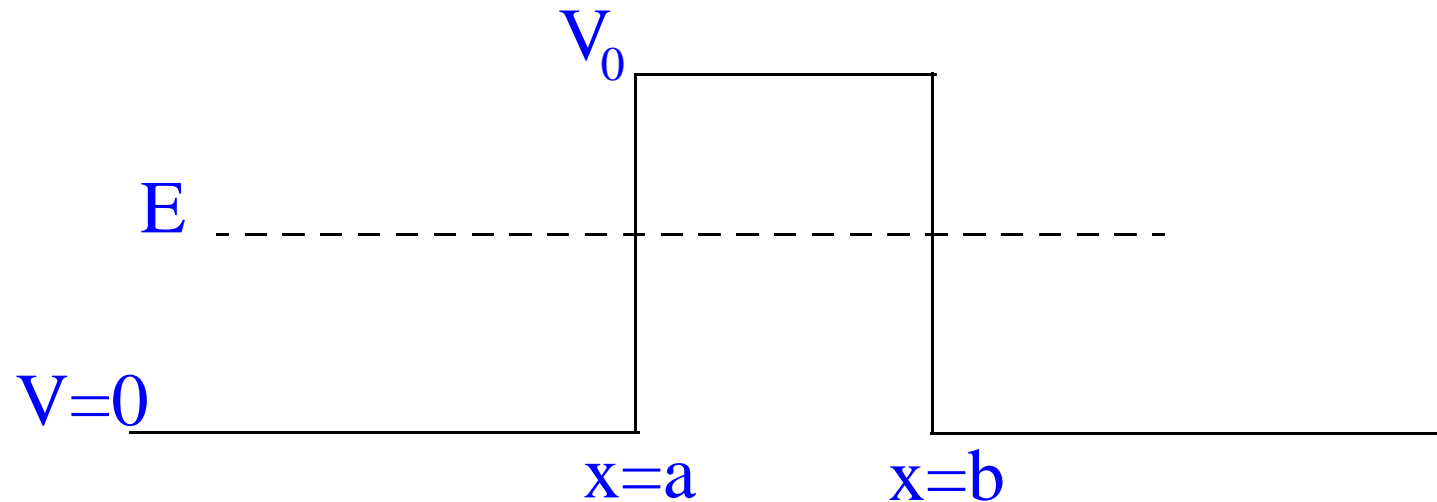
In the frame of the decayed nucleus:

$$Q = K_Y + K_\alpha$$

The kinetic energy of the alpha (a few MeV):

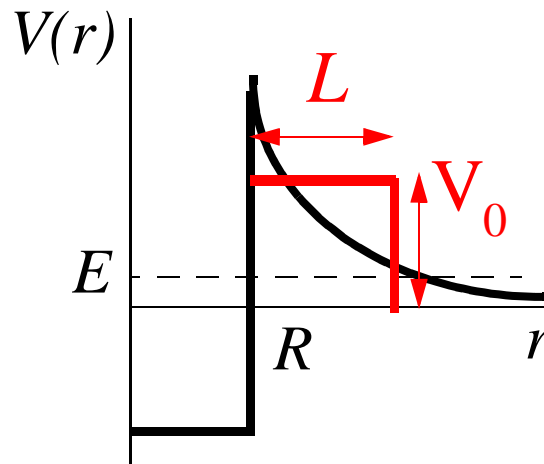
$$K_\alpha \cong \frac{A-4}{A} Q$$

The tunnel effect



A particle with $E < V$ is coming from the left.
What happens when the particle reaches $x=a$?

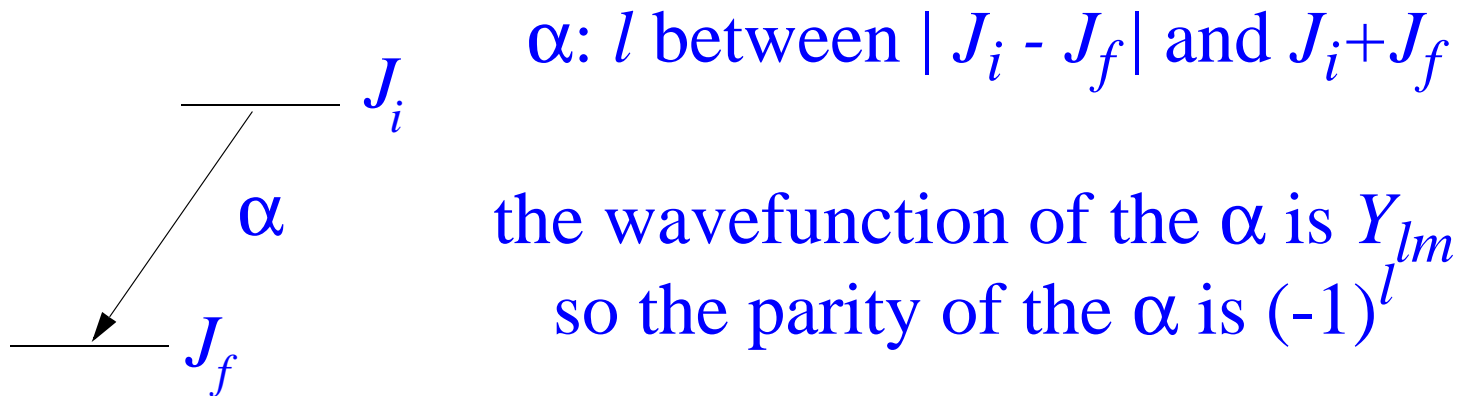
The tunnel effect in α -decay



probability for α -decay:

$$\lambda = \frac{v}{2R} e^{-2kL}, \quad k = \sqrt{2m(2\pi/h)^2 (V_0 - E)}$$

Alpha decay: selection rules (Krane §8.5)



α : l between $|J_i - J_f|$ and $J_i + J_f$

the wavefunction of the α is Y_{lm}
so the parity of the α is $(-1)^l$

Parity is conserved in alpha decay

The following transitions are allowed:

- the initial and final state have same parity and l is even
- the initial and final state have different parity and l is odd

Beta decay: electron emission



by neutron decay $n \rightarrow p + e^- + \bar{\nu}_e$

Q value:

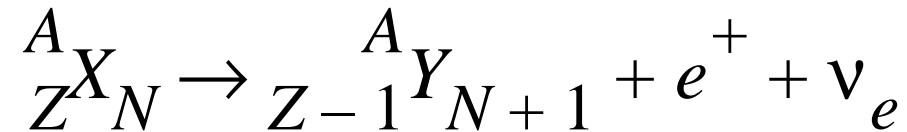
$$Q = m({}^A X) - m({}^A Y) \text{ (atomic masses)}$$

neglecting the recoil energy of the proton, we have

$$Q = K_{\bar{\nu}} + K_{e^-} \text{ (e and } \nu \text{ relativistic)}$$

The kinetic energy of the electron has a continuous spectrum with maximum value $(K_{e^-})_{max} \approx Q$

Beta decay: positron emission



by the transformation $p \rightarrow n + e^+ + \nu_e$
(this is not a proton decay!)

Q value:

$$Q = m({}^A X) - m({}^A Y) - 2m_e \text{ (atomic masses)}$$

As in electron emission, we have $Q = K_\nu + K_{e^+}$ and the kinetic energy of the positron is a continuous distribution with maximum value $(K_{e^+})_{max} \approx Q$

Beta decay: selection rules (Krane §9.4)

We consider the e and ν produced at $r=0$ so that $l=0$

- Fermi decay (e and ν spins are antiparallel, $S=0$):

$$\Delta J_{\text{nucleus}}=0$$

- Gamow-Teller decay (e and ν spins are parallel, $S=1$):

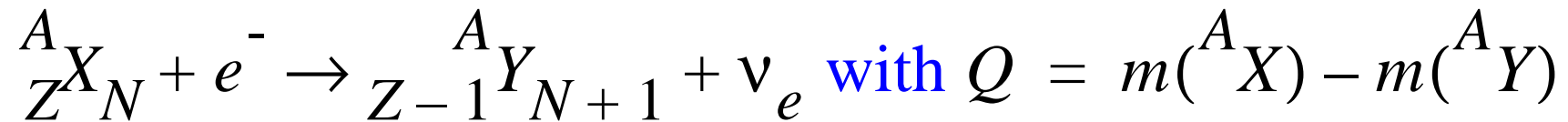
$$\Delta J_{\text{nucleus}}=0, 1 \text{ (except for } J_{\text{initial}}=J_{\text{final}}=0)$$

Since $l=0$ for the electron-neutrino system, we must have

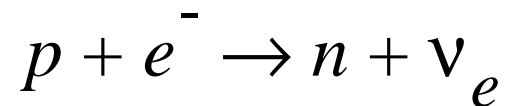
$$\Delta P_{\text{nucleus}}=0$$

‘First-forbidden decays’: $\Delta J_{\text{nucleus}}=0, 1, 2$ and P_{nucleus} changes

Electron capture



by the process



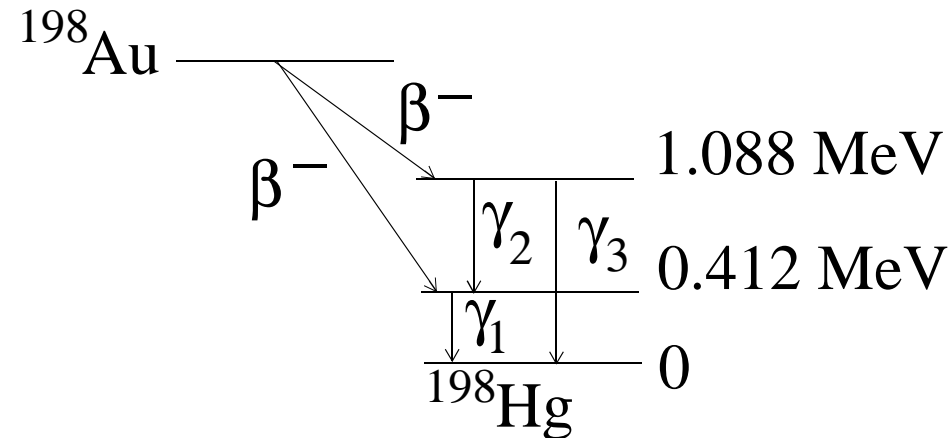
where the electron is an *atomic* electron from a shell with low principal quantum number n

For $n=1$, the electron comes from the K shell and the above process is called 'K-shell capture', etc.

Isomeric transitions

Transitions from an excited nuclear state to a state of lower energy (by photon emission)

Such transition can occur after an α - or β -decay, e.g.



Isomeric transitions: selection rules

A photon in a multipole carries $L(\hbar/2\pi)$
so, in a transition L is between $|J_i - J_f|$ and $J_i + J_f$

The total parity is conserved

- If the parity of the nucleus changes, the radiation field must have odd parity, e.g. E1, E3, ..., or M2, M4, ...
- If the parity of the nucleus does not change, the radiation field must have even parity, e.g. E2, E4, ..., or M1, M3, ...