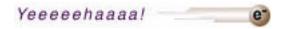
Momentum eigenstates

- **T** linear momentum eigenstates
- Torbital angular momentum eigenstates
- spin angular momentum eigenstates
- **T** addition of angular momenta

→ the generic angular momentum operator

We will work mainly with quantum numbers in the applications



Linear momentum eigenstates

We will consider motion in one dimension (x):

$$\boldsymbol{p}_{\boldsymbol{x}} = -i\frac{h}{2\pi}\frac{\mathrm{d}}{\mathrm{d}x}$$

By solving the eigenvalue equation $p_x \psi = p_x \psi$ (where both ψ and p_x are unknown), we find

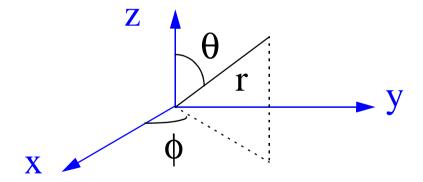
$$\psi(x) = c \cdot e^{\left(\frac{2\pi i}{h}\right)xp_{x}}$$

Eigenstates of orbital angular momentum

We will look for the simultaneous eigenstates ψ of the operators \boldsymbol{L}^2 and \boldsymbol{L}_z

$$\boldsymbol{L}_{z}\psi = L_{z}\psi, \quad \boldsymbol{L}^{2}\psi = L^{2}\psi$$

We will use spherical coordinates: (r, θ, ϕ)



L_z eigenstates and eigenvalues

The eigenvalue equation

$$-i\frac{h}{2\pi}\frac{\mathrm{d}\psi}{\mathrm{d}\phi} = L_z \text{ gives}$$

$$\psi(\phi) = ce^{\left(\frac{2\pi i}{h}\right)\phi L_z}$$

quantized

Condition:
$$\psi(\phi) = \psi(\phi + 2\pi) \Rightarrow L_z = m\frac{h}{2\pi}$$

 $m = 0, \pm 1, \pm 2, \dots$ is the magnetic quantum number

Eigenstates of L^2 and L_z

Spherical harmonics

$$Y_{lm}(\theta, \phi) = (-1)^m \left[\frac{(2l+1)(l-m)!}{4\pi(l+m)!} \right]^{1/2} P_l^m(\cos\theta) e^{im\phi}$$

where $P_l^m(\cos\theta)$ are the associated Legendre functions

$$L^2 = \left(\frac{h}{2\pi}\right)^2 l(l+1), \ l = 0, 1, 2, \dots$$
 the orbital quantum number

the magnetic quantum number is restricted by l:

$$m = -l, -l + 1, ..., 0, ..., l - 1, l$$

Parity eigenstates

The spherical harmonics are also eigenstates of the parity operator

$$PY_{lm} = (-1)^{l} Y_{lm}$$

Application: transitions in atoms

The generic angular momentum operator

$$J = J_x \overrightarrow{u}_x + J_y \overrightarrow{u}_y + J_z \overrightarrow{u}_z$$
, $[J_x, J_y] = i \frac{h}{2\pi} J_z$ and cyclic permutations

$$J^{2} = \left(\frac{h}{2\pi}\right)^{2} j(j+1), L_{z} = m\frac{h}{2\pi}, j = 0, 1, 2, ..., m_{j} = -j, -j+1, ..., j$$

By solving the eigenvalue equations for J^2 and J_z simultaneously and assuming n eigenstates, we find that there are two 'types' of angular momentum:

- \checkmark integer j
- \checkmark half-integer j

The spin angular momentum

Spin does not involve a rotation in space

$$S^2 = \left(\frac{h}{2\pi}\right)^2 s(s+1)$$

where s is the spin quantum number

$$S_z = m_s \frac{h}{2\pi}, \qquad m_s = -s, -s+1, \dots, s-1, s$$

- ✓ integer s: bosons
- ✓ half-integer s: fermions

Addition of angular momenta

Let us assume that we want to 'add' the quantum numbers l and s to find the quantum number of the total angular momentum, j, and the quantum number of the z-component, m_j

The rules are:

- 1. j = |l s|, ..., l + s
- 2. for each j, we have 2j + 1 m_j 's: $m_j = -j, -j + 1, ..., j$

Application: L-S coupling and J-J coupling in atoms